

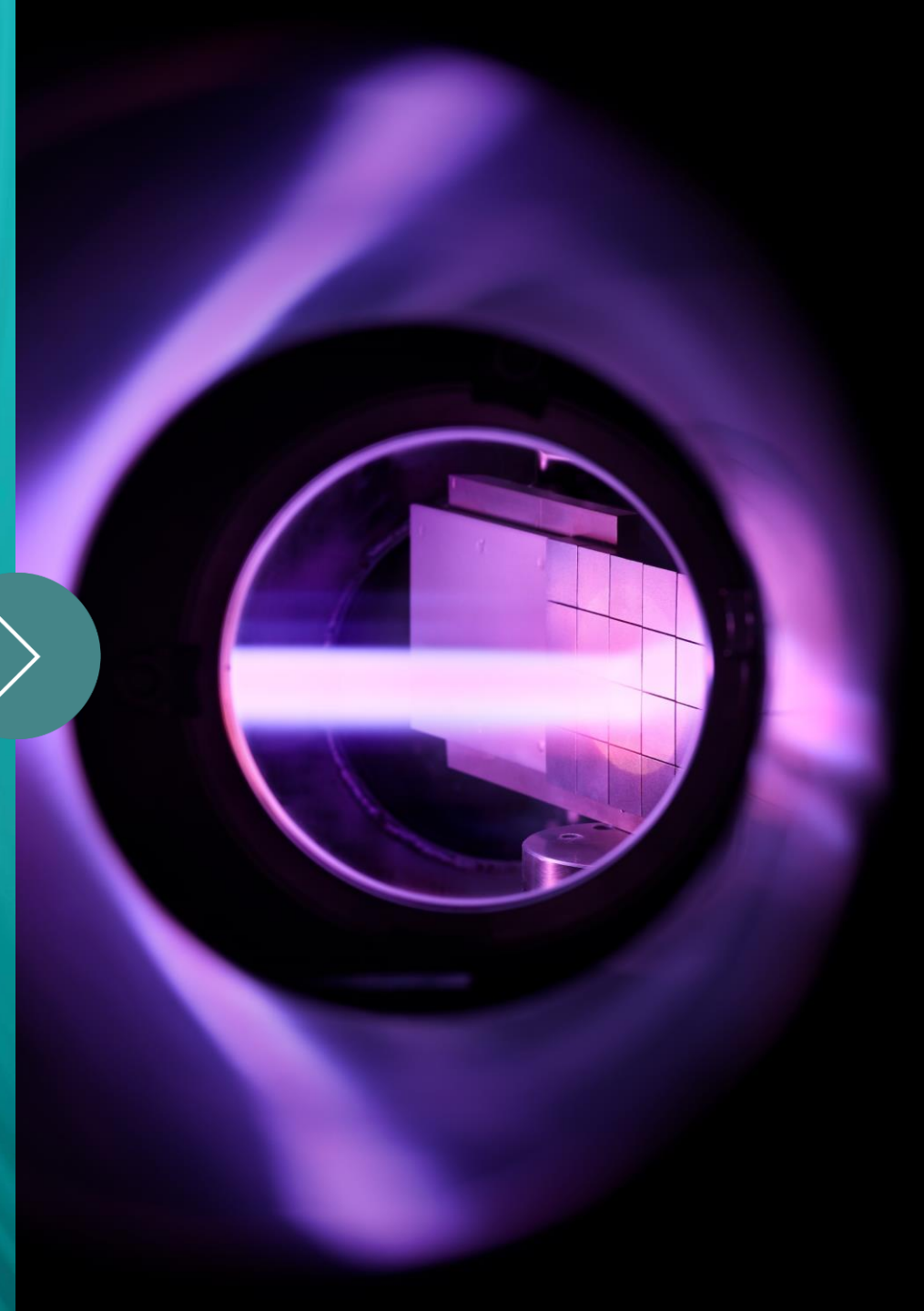
SPA 3 Activities DIFFER 2023

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UHTC Overview/Background

- **Ultra-High Temperature Ceramics (UHTC)**
- Defined by:
 - Melting temperatures: > 3000 °C
 - Application: Sustained working temperatures > 1600 °C
 - Chemistry: Largely binary compounds of Boron, Nitrogen, or Carbon bonded with an early transition metal
- Historic and current materials of interest for leading edges, heat shields, and thermal protection systems for hypersonic and atmospheric re-entry vehicles

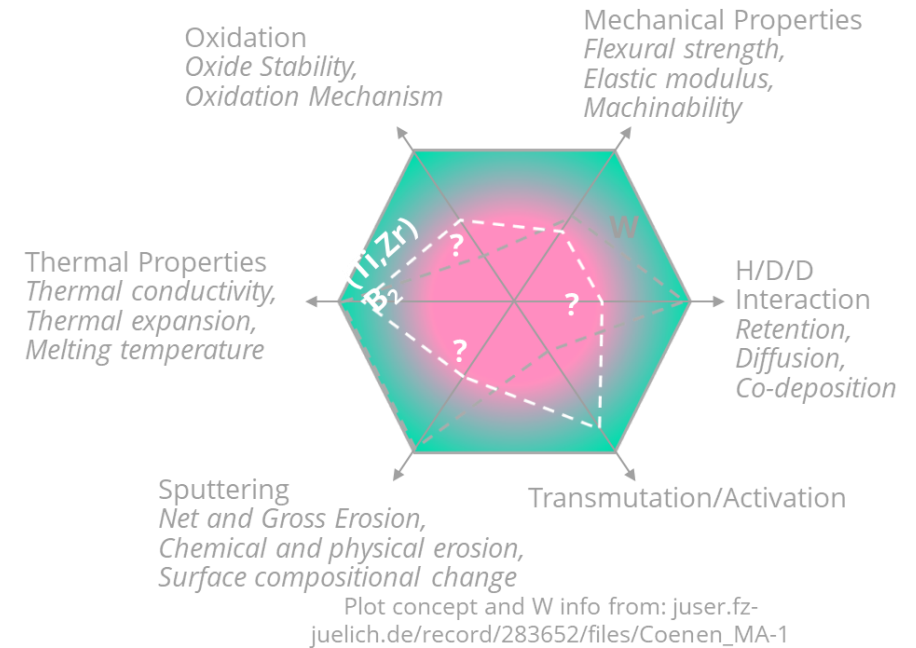
PubChem

Hydrogen Nonmetal		Name Chemical Group Block															
1 H Hydrogen	2 He Helium	3 Li Lithium	4 Be Beryllium	5 B Boron	6 C Carbon	7 N Nitrogen	8 O Oxygen	9 F Fluorine	10 Ne Neon	11 Na Sodium	12 Mg Magnesium	13 Al Aluminum	14 Si Silicon	15 P Phosphorus	16 S Sulfur	17 Cl Chlorine	18 Ar Argon
19 K Potassium	20 Ca Calcium	21 Sc Scandium	22 Ti Titanium	23 V Vanadium	24 Cr Chromium	25 Mn Manganese	26 Fe Iron	27 Co Cobalt	28 Ni Nickel	29 Cu Copper	30 Zn Zinc	31 Ga Gallium	32 Ge Germanium	33 As Arsenic	34 Se Selenium	35 Br Bromine	36 Kr Krypton
37 Rb Rubidium	38 Sr Strontium	39 Y Yttrium	40 Zr Zirconium	41 Nb Niobium	42 Mo Molybdenum	43 Tc Technetium	44 Ru Ruthenium	45 Rh Rhodium	46 Pd Palladium	47 Ag Silver	48 Cd Cadmium	49 In Indium	50 Sn Tin	51 Sb Antimony	52 Te Tellurium	53 I Iodine	54 Xe Xenon
55 Cs Cesium	56 Ba Barium	•	72 Hf Hafnium	73 Ta Tantalum	74 W Tungsten	75 Re Rhenium	76 Os Osmium	77 Ir Iridium	78 Pt Platinum	79 Au Gold	80 Hg Mercury	81 Tl Thallium	82 Pb Lead	83 Bi Bismuth	84 Po Polonium	85 At Astatine	86 Rn Radon
87 Fr Francium	88 Ra Radium	•	104 Rf Rutherfordium	105 Db Dubnium	106 Sg Seaborgium	107 Bh Bohrium	108 Hs Hassium	109 Mt Meitnerium	110 Ds Darmstadtium	111 Rg Roentgenium	112 Cn Copernicium	113 Nh Nihonium	114 Fl Flerovium	115 Mc Moscovium	116 Lv Livermorium	117 Ts Tennessine	118 Og Oganesson
•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•	•
•	•	•	57 La Lanthanum	58 Ce Cerium	59 Pr Praseodymium	60 Nd Neodymium	61 Pm Promethium	62 Sm Samarium	63 Eu Europium	64 Gd Gadolinium	65 Tb Terbium	66 Dy Dysprosium	67 Ho Holmium	68 Er Erbium	69 Tm Thulium	70 Yb Ytterbium	71 Lu Lutetium
•	•	•	89 Ac Actinium	90 Th Thorium	91 Pa Protactinium	92 U Uranium	93 Np Neptunium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	97 Bk Berkelium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium	103 Lr Lawrencium



UHTCs Material Properties and Fusion Plasma Facing Applications

- Several UHTC chemistries have appealing material properties for fusion PFM applications
 - High temperature strength and thermal conductivity
 - tunable neutronic properties
 - potentially composed of low to mid Z elements
- However, uncertainties on UHTC PMI response needs to be studied prior to qualification/disqualification
 - chemical erosion processes
 - failure mechanisms at high temperatures
 - microstructural response to plasma transients

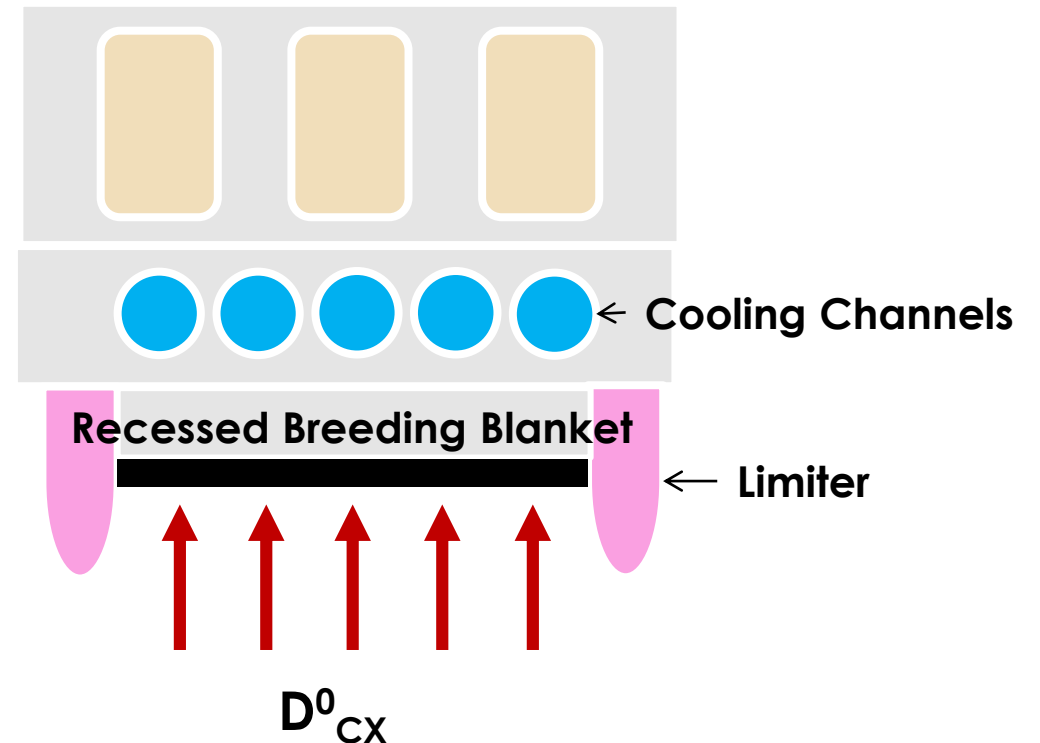


Material	Melting Temperature (°C)	RT Flexural Strength (MPa)	HT Flexural Strength (MPa)	RT Thermal Conductivity (W/m-K)	HT (1000 °C) Thermal Conductivity (W/m-K)
ZrB ₂	~3245	300 - 500	~450 (800 °C)	60 - 105	~50
TiB ₂	~3225	375 - 1000	~550 (1000 °C)	60 - 120	~67
Sintered SiC	2700 (sublimation)	325 - 400	350 - 450 (1000 °C)	~400	~80
Tungsten	~3422	460-600	~200 (1000 °C)	~180	~110



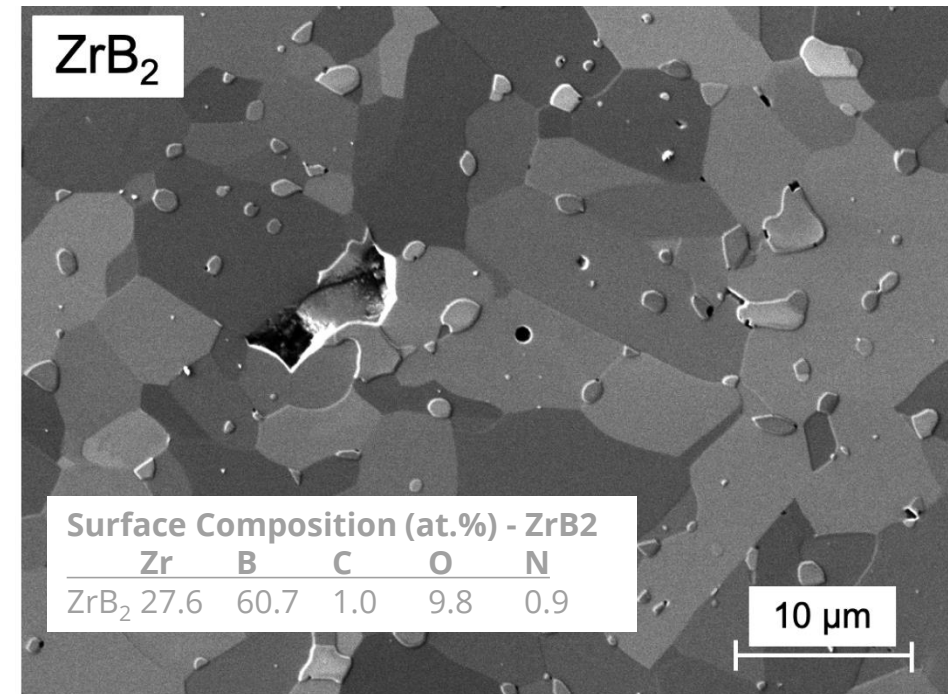
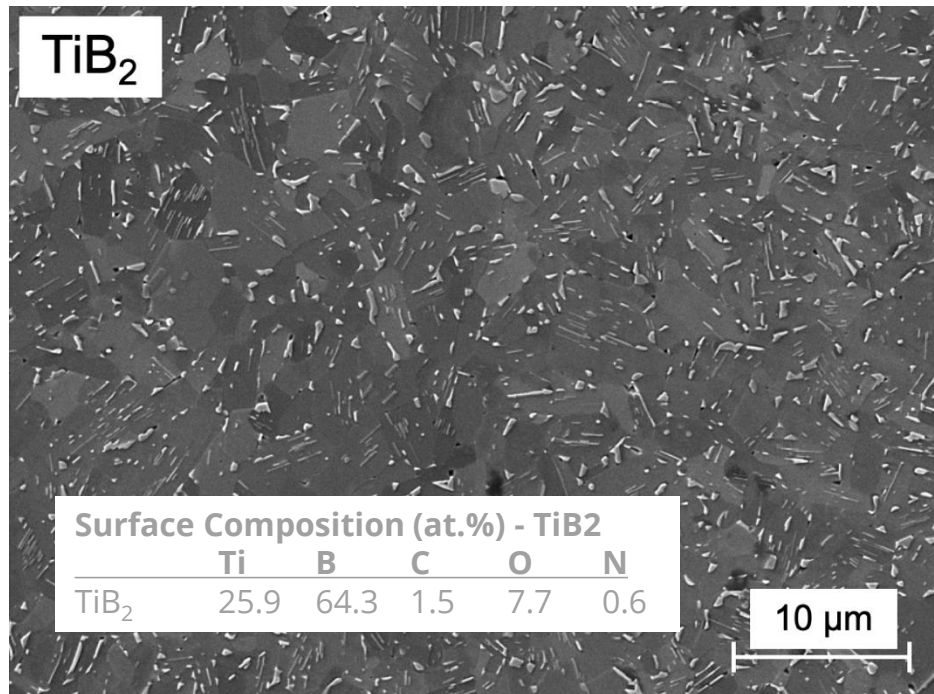
UHTCs for Sacrificial Limiters

- Moving towards a fusion pilot plant (FPP) means more demands on first wall (FW) blanket
 - Heat transfer and tritium breeding requirements verses just surviving
 - Stricter heat flux constraints
- Sacrificial limiters will likely be used to protect the blanket from transients
 - Limiter Drivers:
 - 1) Withstand thermal and particle loads from steady state exposures and ramp-up and ramp-down
 - 2) Following intense/unexpected transients, protect the blanket from damage extending to cooling channels



Starting Materials

- ZrB_2 and TiB_2 samples formed via reaction hot pressing
 - $\text{MeH}_2 + 2\text{B} \rightarrow \text{MeB}_2 + \text{H}_2(\text{gas})$
 - No sintering additives used to ensure chemical purity
 - Pristine grain sizes $\sim 3 - 10 \mu\text{m}$ (TiB_2) and $\sim 10 - 30 \mu\text{m}$ (ZrB_2)



SEM micrographs taken with Tescan Mira 3 at LAMDA labs (ORNL), surface chemistry measurements taken via x-ray photoelectron spectroscopy.



Proposed Magnum PSI Exposure Conditions

- Experimental Goals:
 - **1) Analyze material surface microstructure evolution as a response to coupled steady-state and transient heat fluxes**
 - **2) Determine transient load limit in diborides**
 - **3) Measure erosion products in-situ and net erosion ex-situ to determine changes in erosion behavior as a response to coupled HF**
- Utilizes the pulsed plasma system in tandem with steady-state exposures
- Experimental Controls
 - Plasma composition
 - Steady-state loading conditions (~5 MW/m², ion flux, ion energy, ion fluence)
 - Transient pulse conditions (~10 Hz, 0.25 GW/m², 1 ms pulses)
- Experimental Variables
 - Presence of pulses
 - Number of total pulses





**Thank you for your
attention**



Preliminary Shot Plan

Sample Name	Chemistry	# of transient pulses	Steady State Heat Flux (MW-m ⁻²)
TB-1	TiB ₂	0	5
TB-2	TiB ₂	10	5
TB-3	TiB ₂	100	5
TB-4	TiB ₂	1000	5
TB-5	TiB ₂	10000	5
TB-A*	TiB ₂	0	5
TB-B*	TiB ₂	100	5
ZB-1	ZrB ₂	0	5
ZB-2	ZrB ₂	10	5
ZB-3	ZrB ₂	100	5
ZB-4	ZrB ₂	1000	5
ZB-5	ZrB ₂	10000	5
ZB-A*	ZrB ₂	0	5
ZB-B*	ZrB ₂	100	5

*D plasma exposures with transient heat flux simulation from LASGAG system

Desired in-situ diagnostics

- Fast IR imaging of sample surfaces
- Thomson scattering on plasma near target surface
- OES tuned to target elements

Controls (excluding *'d samples)

- H plasma
- **Pulse magnitude (0.25 GW/m²), frequency (10 Hz) and duration (1 ms)**
- Steady state conditions
 - HF, ion energy, ion flux, fluence
 - 1000 °C ambient surface temperature

Planned Post-mortem Analysis

- IBA-NRA
- Mass loss measurements
 - Net erosion
- X-ray photoelectron spectroscopy (XPS)*
 - Measures changes in areal-averaged surface chemistry/stoichiometry, indicating preferential erosion
- Scanning electron microscopy (SEM)*
 - Examine plasma-induced microstructural evolution of sample surfaces
- SEM-based techniques: energy dispersive x-ray spectroscopy (EDS)* and electron backscattered diffraction (EBSD)*
 - EDS: Surface chemistry mapping to determine the presence and magnitude of preferential erosion
 - EBSD: Surface crystallographic mapping to measure changes crystallographic orientation
- Focus ion beam (FIB)*
 - Determine damage depth
- Transient grating spectroscopy (TGS)
 - Measures surface thermal transport properties
- Thermal desorption spectroscopy (TDS)*
 - Measures light atom retention in material
- Atomic Force Microscopy (AFM)*
 - Measures/maps surface roughness